# SBG Study: Download Free Study Material WWW.SBGSTUDY.COM

**TME-302** 

Roll No.

(ODD SEM) Examination 2019-20 B.TECH. Semester III<sup>rd</sup> MECHANICAL/AUTO/PLASTIC & POLYMER

**Engineering Thermodynamics** 

Time: 2 Hrs.

**Total Marks: 50** 

Total no. of printed pages: 1

Note: Attempt all questions.

#### Q.1 Answer any three of the following:

 $(4 \times 3)$ 

- (a) Give the concept of first law of thermodynamics. What are its limitations?
- (b) Explain steady flow energy equation. Apply this equation for adiabatic expansion of steam through nozzle and compression of air through a compressor
- (c) Define adiabatic process and develops an expression for work done by a gas during its adiabatic expansion.
- (d) Define availability and develop an expression for maximum work developed in a reversible process.

## Q.2 Answer any three of the followings:

 $(4 \times 3)$ 

- (a) Explain joule Kelvin effect with a neat sketch of throttling process. Prove that Joule Thomson coefficient  $\mu = 0$  for an ideal gas.
- (b) What do you understand by high grade energy and low grade energy? Give any four examples of each type.
- (c) What is condition for exact differential? Derive Maxwell's Equation.
- (d) Calculate the available energy in 80 kg of water at 50°C with respect to surrounding at 5°C. The pressure of water being 1 atm.

### Q.3 Attempt any two of the following:

 $(6.5 \times 2)$ 

- (a) Explain Joule-Kelvin effect? What is inversion curve, show it on T-P diagram.
- (b) Define  $\acute{\eta}$  of a reversible heat engine. Draw Carnot cycle on (P-V) & (T-S) diagram and develop an expression for  $\acute{\eta}$  of Carnot heat engine cycle
- (c) Define available and unavailable energy. Explain available energy referred to a cycle.

#### Q.4 Answer any two of the following:

 $(6.5 \times 2)$ 

- (a) Compare Otto, Diesel and Dual cycles on the basis of same compression ratio and heat input. Draw cycles on P-V and T-S diagram.
- (b)  $m_1$  kg of water at  $T_1$  is isobarically and adiabatically mixed with  $m_2$  kg of water at  $T_2$ , where  $T_1 > T_2$ . Show that for equal masses of water, entropy change of mixture is given by

$$dS(universe) = 2 mc_p \log_e \left( \frac{T_1 + T_2}{\sqrt{T_1 \cdot T_2}} \right)$$

(c) Discuss the working of a centrifugal compressor with a neat sketch. What is difference in principle of working of it and axial flow compressor?