

SBG STUDY

Bits

CMP - CH 15 50/10

J-A - JM

~~RAO - X~~

06/09/17

chapter :

Modern Physics

* Photoelectric effect :

Light is constitute of small particles known as photons

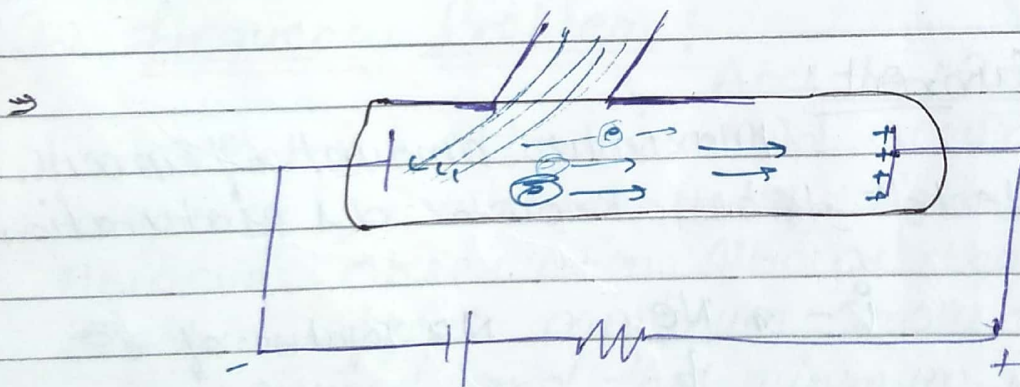
* Photons have Rest mass = 0

* Photons have momentum = $p = \frac{h}{\lambda}$

$$p = \frac{E}{c}$$

* Photons have Energy $E = \frac{hc}{\lambda}$

* Photoelectric effect shows that light have Particle nature and its particles are known as photons



* Some Important Definitions?

1) Work function : minimum amount of energy required for the removal of electron from metal is known as work function.

Work function is minimum for Cesium and
It is relatively for Alkali metals.

(2) Photo electrons!

Electrons Removed from metal
on the incidence of light is known as Photo-
electrons

(3) Photo current:

Current due to flow of
these e^- (photo electrons) is known as Photo
Current

$$V = IR \rightarrow X$$

$$i = \frac{Q}{t} = \frac{ne}{t}$$

$n = \text{no. of photo } e^-$

* Saturation Current:

Maximum amount of current
in a photo electric tube known as Saturation
Current.

$$i = \frac{Ne}{t}$$

$N \Rightarrow \text{Total no. of } e^-$

* Stopping Potential:

Reverse amount of Potential
applied across photo tube, so that none of
the electron can reach other plate is known
as Stopping potential

$$K E_{\text{max}} = e V_s$$

$V_s = \text{Stopping Potential}$

Observation made by Einstein:

(1) Time delay:

(1) Intensity Problem:

Int. \propto Electric f.

Acc. to wave theory Intensity of light depends on amplitude of electric field as amplitude / as intensity of light will increase and amplitude of light will increase which result will increase force on electron.

Due to which kinetic energy of e^- must increase. And stopping potential also increase. However observation shows that maximum K.E is independent from Intensity of light.

(2) Frequency Problem:

Acc. to wave theory Photoelectric effect should occur for any frequency (wave length of light) provided. However observation shows that for photoelectric effect minimum amount of Energy is required and this minimum amount of frequency is known as Threshold frequency and it was observed that photoelectric effect shown as light of wave length less than a particular maximum wave length known as Threshold wave length.

$E = h\nu_0$ Threshold frequency

Board

$$\text{Threshold Energy} = h\nu_0$$

(3) Time delay:

There was no time delay b/w emission of photo electron and incidence of light which proves that light has particle nature

* Einstein eqⁿ for photo electric effect:

Acc. to photon theory of Einstein's energy of ~~photo~~ light is quantised and the small packets of energy are known as photons

$$\text{Energy carried by one photon} = \frac{hc}{\lambda}$$

$$I = nh\nu$$

$$\nu = \frac{c}{\lambda}$$

I = Energy per unit area per unit time.

$$I = nh\nu \quad \text{or} \quad h\nu$$

where n is no. of photons per unit area or per unit time.

$$E = h\nu_0$$

ν_0 = Threshold frequency

Qmb

$$E = \phi_0 + K.E$$

$$h\nu = \phi_0 + K.E_{\text{max}}$$

$$\nu \geq \nu_0$$

↳ Threshold Energy

18 1.6

$$E = \frac{123.400}{\lambda \text{ in } \text{Å}} = \text{eV.}$$

$$\frac{hc}{\lambda} = \frac{1.6 \times 10^{-19}}{\lambda}$$

$$h = 6.6 \times 10^{-34}$$

$$c = 3 \times 10^8$$

$$\frac{1.6 \times 10^{-19}}{\lambda} \text{ for convert in eV.}$$

Que! A wave length

Light having wave length of 4000 Å falls on metal surface having work function 2 eV . find max. kinetic energy. also find stopping potential

$$h\nu = \phi_0 + K.E. \text{ max.}$$

~~$$K.E. = h\nu - \phi_0$$~~

$$K.E. \text{ max} = h\nu - \phi$$

$$= \frac{hc}{\lambda} - \phi$$

$$= \frac{1.6 \times 10^{-19}}{4000 \times 10^{-10}} - 2$$

$$= \frac{1.6 \times 10^{-19}}{4000 \times 10^{-10}} - 2$$

1.6

$$h = \frac{6.6 \times 10^{-34} \times 3 \times 10^8}{\lambda \times 10^{-10}}$$

$$= \frac{19.2 \times 10^{-17}}{\lambda}$$

Ans! $E = \frac{hc}{\lambda} = \frac{12.400}{400} = 3.1 \text{ eV.}$

$$E = \phi + K.E. \text{ max}$$

$$= \frac{1.1 \text{ eV}}{1.1 \text{ V}}$$

$$3.1 = 2 \text{ eV} + K.E. \text{ max}$$

$$1.1 \text{ eV} = K.E. \text{ max} = eV_s$$

⇒

$$1.6 \times 10^{-19} \text{ J} = 1 \text{ eV.}$$

$$= \frac{19.2 \times 10^{-16}}{A [1.6 \times 10^{-19}]} = \frac{19.2 \times 10^3}{A \cdot 1.6}$$

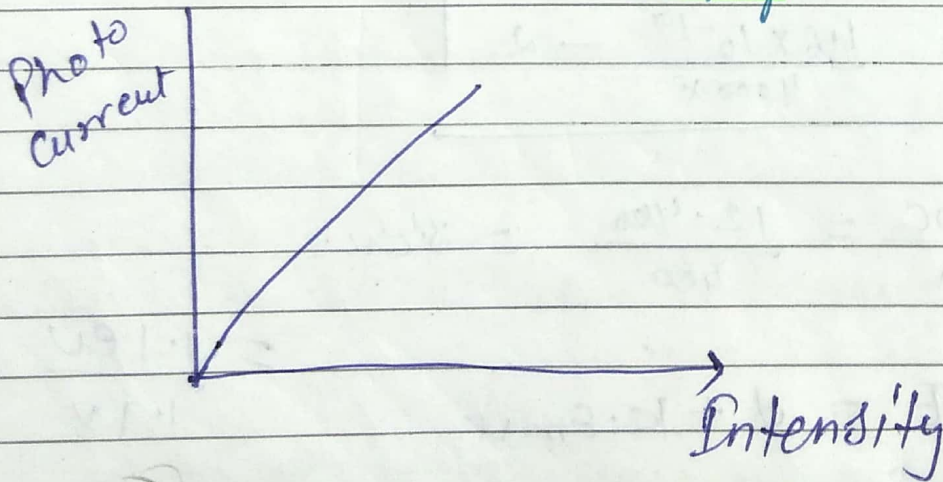
$$= \frac{12340}{A} \Rightarrow$$

where A is any no. in \AA° .

$$E = \frac{hc}{\lambda} = \frac{12340}{\lambda \text{ in } \text{\AA}}$$

* Some Important terms / graphs / observation :-
for photo electric effect:

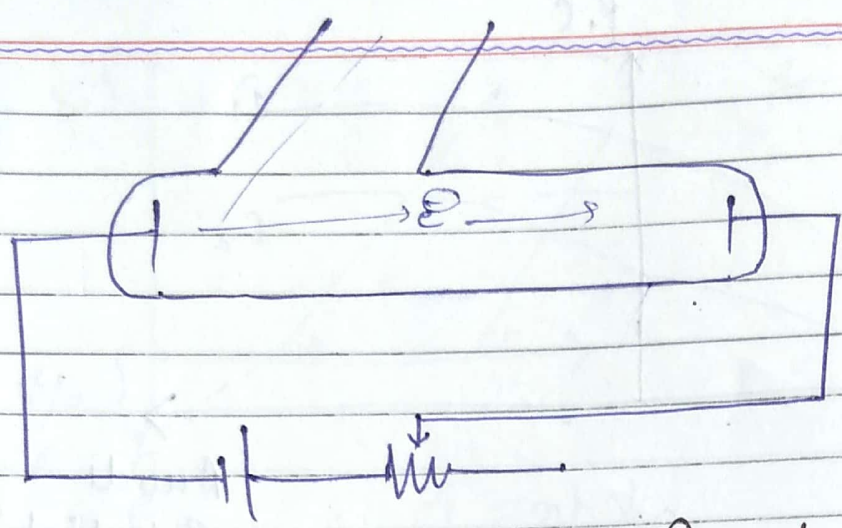
(1) Variations of Photo Current with the various
of Intensity ~~varying~~ keeping frequency const.



$$I = nh\nu$$

where $I = \frac{ne}{t}$
 $I \propto n$
Intensity \propto Probability

λ - frequency



Einstein Photon theory or Quantum theory :

$$E = \frac{nhc}{\lambda} = nh\nu$$

According to Einstein Energy of light is Quantised in form of photons which behaves as particles. When photons collides with electron total energy of a photon will be absorbed by electrons (as this energy cannot be further divided).

not very clear part

$$I = \frac{ne}{t}$$

$$I \uparrow = n \uparrow$$

voltage \uparrow current \uparrow
 $n \uparrow$
 ($n=0$) ($I=0$ current)

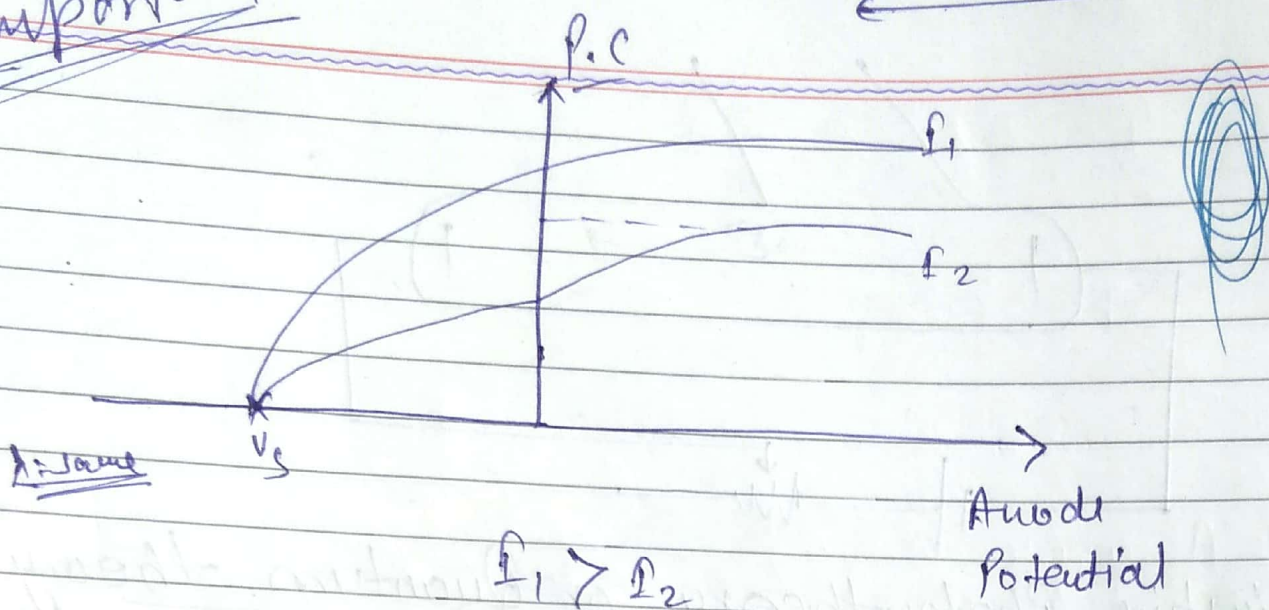
9) Variation of photo current with Potential difference applied across tube

(A) Keeping frequency constant.

colour fixed.

~~Important~~

VIBRYOR

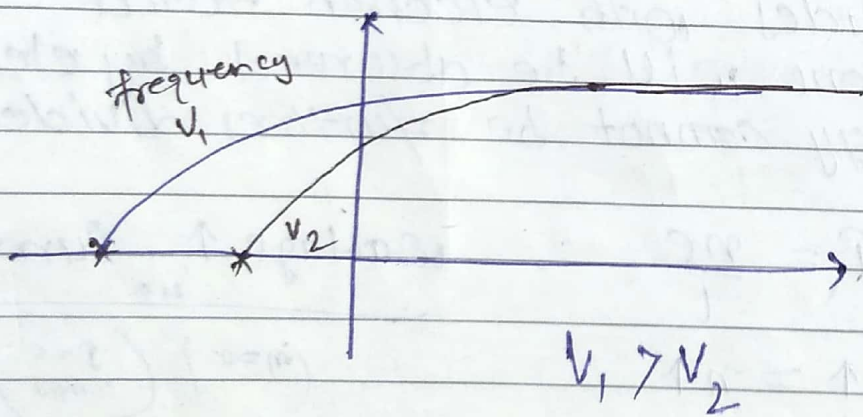


(B) Keeping Intensity Constant.

Changing frequency

n same

$$E = \frac{nhc}{\lambda}$$



* Variation of Stopping potential with frequency:

$$h\nu = \phi_0 + eV_s$$

$$V_s = \frac{h\nu}{e} - \frac{\phi_0}{e}$$

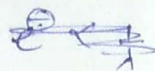
$$V_s = \text{slope} \times \nu + c$$

slope

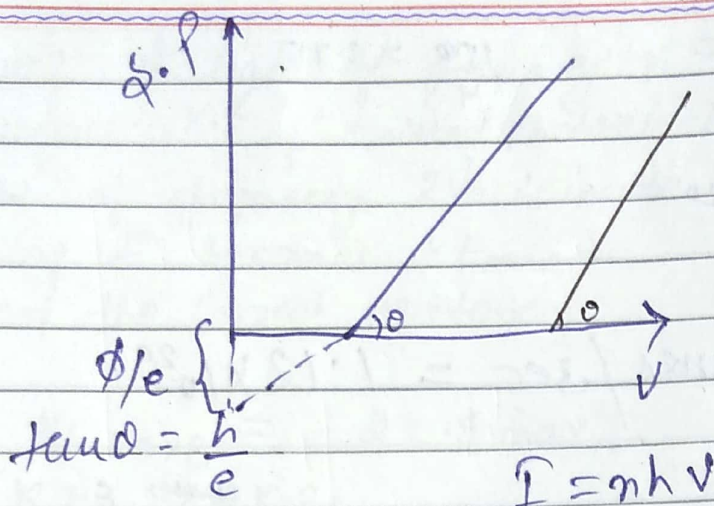
$KE = eV_s$

$$1.6 \times 10^{-19}$$

Remove



19



for any metal.
slope is not change

Remove \times speed
minimum freq.
threshold

Que! A Yellow light having wavelength 5000 \AA is used for photo electric effect source having wattage of ~~1000~~ 40 watt is used

Power

find

- (i) no. of photons emitted by source
- (ii) no. of photons which will strike metallic plate of Area 1 m^2
- (iii) if work fn of metal is 2 eV . find maximum kinetic energy of e^- and stopping potential V
- (iv) if efficiency of emission is 80%. find saturation current.

$$E = \frac{n h c}{\lambda} \quad n = \frac{E \lambda}{h c} = \frac{E \cdot 5000}{12340}$$

$$E = \frac{h c}{\lambda} = \frac{12340}{5000}$$

$$E = \frac{n h c}{\lambda} = \frac{12.400}{5.00} = 2.48 \text{ eV}$$

$$I = 40 \frac{\text{Joule}}{\text{sec}} = n [2.48 \times 1.6 \times 10^{-19} \text{ Joule}]$$

$$\frac{40}{1.6 \times 2.48} \times 10^{19} = n$$

$$h\nu = \phi_0 + eV_s.$$

0-1
photo electric effⁿ
0-2 - EME

$$3.6 \frac{2.48 \times 1.6}{4} = \frac{1500}{4} = 3.75$$

$$1.12 \times 10^{20}$$

$$\text{no. of photons/sec} = 1.12 \times 10^{20}$$

(ii)

$$\frac{P}{\text{sec}} = \text{watt}$$

$$\text{watt} = \epsilon$$

$$\frac{1.12 \times 10^{20}}{\text{m}^2 \text{ sec}}$$

no. of photo
e⁻/sec

$$\frac{nhc}{\lambda}$$

$$= 0.56 \times 10^{20}$$

$$\text{or } 5.6 \times 10^{19}$$

(3)

$$5.6 \times 10^{19}$$

Current

(iv)

80% efficiency \rightarrow 80 e⁻ Removed.

$$5.6 \times 10^{19} \times 0.8$$

$$= 4.48 \times 10^{18} \text{ e}^-$$

$$i = \frac{ne}{t}$$

$$= 4.48 \times 1.6 \times 10^{-19}$$

$$i = 0.7 \text{ A}$$

* Que: A light of frequency ν_0 is incident on a metal surface K.E of emitted photo-electron is KE_{maxA} if light of frequency $2\nu_0$ is incident on same metal surface. K.E of e^- becomes KE_{maxB} .
 select the correct statement.

(A) $KE_{maxB} = 2 KE_{maxA}$

~~(B) $KE_B > 2 KE_A$~~

(C) $KE_B < 2 KE_A$

(D) None.

$$E = \phi + KE$$

$$\downarrow \quad \downarrow$$

$$h\nu \quad 2 + KE$$

$$e = 2 + KE$$

Que: Light of wavelength $1.98 \times 10^{-7} \text{ m}$ is incident on a metal surface it has been observed that stopping potential is 2.5 Volt find

(i) Maximum speed of the photo electron, work function and threshold frequency.

$$h\nu = \phi + KE$$

$$2.5 \text{ V} = \frac{1.98 \times 10^{-7}}{1.6 \times 10^{-19}} - \phi$$

Ans!

$$2.5 \text{ eV} = \phi$$

$$2.5 \times 1.6 \times 10^{-19} = \frac{1}{2} mv^2$$

$$2.5 \times 1.6 \times 10^{-19} = \frac{1}{2} \times 9 \times 10^{-31} \times v^2$$

$\frac{V_{st}}{e}$ stopping potential

$$\frac{8}{9} \times 10^{12} = v^2$$

0.99

$$10^6 = v$$

(ii) $h\nu = \phi_0 + K.E$

$$h\nu = \phi_0 + 2.5 \text{ eV}$$

$$\frac{12400}{1980} = \phi + 2.5 \text{ eV}$$

$$6.2 = \phi + 2.5$$

$$6.2 - 2.5 = \phi$$

$$\phi = 3.76 \text{ eV}$$

$$\phi = \frac{3.7 \times 1.6 \times 10^{-19}}{6.6 \times 10^{-34}} = 9.1 \times 10^{14}$$

single electron

Ques: A metallic surface is radiated with monochromatic light of variable wave length. ^{below} a wave length of 5000 \AA . No. of photo e^- s are emitted from the surface. with an unknown stopping potential of 3 volt is obtained. find wave length of unknown source.

$$= eV = K.E_{\text{max}}$$

$$K.E =$$

Ques:

$$\lambda = 5000 \text{ \AA}$$

$$E = \frac{hc}{\lambda}$$

$$\frac{nhc}{\lambda} = \text{work}$$

Threshold

Solⁿ: $E_{\text{light}} = \phi_{\text{metal}} + K.E$

$$\frac{12400}{\lambda} = \frac{12400}{5.940} + 3 \text{ eV}$$

$$\frac{12400}{5.5} = \lambda$$

Ques

Eliminating of the surface of ^{certain} metal alternately with light of same length $\lambda_1 = 0.35 \mu\text{m}$, $\lambda_2 = 0.54 \mu\text{m}$ it was found at corresponding max velocity of photo e⁻ have a ratio 2:1 find Threshold energy of metal.

Solⁿ:

$$E_1 = \phi + K.E_{\text{max}}$$

$$K.E = \frac{1}{2}mv^2 = \frac{1}{2} \times 1.6 \times 10^{-19} \times v^2$$

$$\left[\begin{array}{l} \frac{12400}{0.35} = \phi + 2 \\ \frac{12400}{0.54} = \phi + 1 \end{array} \right]$$

$$\frac{12400}{0.35} - 2 = \phi$$

$$\frac{12400}{0.54} - 1 = \phi$$

A^o
or
10
then
then
12400

on Next page

use λ in A°

plate or split

3-1 Phot. Ele. Eff. ~~to~~

Sol: $E_{\text{right}} = \text{metal} + k \cdot \epsilon$

$$E_{\text{right}} = \text{Metal} = k \cdot \epsilon$$

$$\epsilon_1 - \phi = k \cdot \epsilon_1$$

$$\epsilon_2 - \phi = k \cdot \epsilon_2$$

$$\frac{\frac{12400}{3500} - \phi_0}{\frac{12400}{5400} - \phi_0} = \frac{4}{1}$$

$$\frac{3.7 - \phi_0}{2.3 - \phi_0} = \frac{4}{1}$$

$$3.7 - \phi_0 = 9.2 - 4\phi_0$$

$$3\phi_0 = 5.5$$

$$\phi_0 = \frac{5.5}{3} = \frac{55}{30} = 1.$$

(checked calculation)

$$I_{\text{light}} = \phi_{\text{metal}} + K E f \quad f = \frac{ne}{t}$$

Last Year
in chem

Ques: A beam of light consists of four wave lengths
 4000 \AA , 4800 \AA , 6000 \AA , 7000 \AA
 each of the intensity $I = 1.5 \times 10^{-3} \text{ watt/m}^2$
 The beams falls normally on an
 Area 10^{-4} m^2 of a metal having work
 function 1.9 eV . Calculate no. of photons emitted
 Also find saturation current.

Solⁿ

$$E_1 = \frac{hc}{\lambda} = E_1 = \frac{12400}{4000} = 3.1 \checkmark$$

$$E_2 = \frac{12400}{4800 \text{ \AA}} = \frac{31}{12} = 2.58 = 2.6 \checkmark$$

$$E_3 = \frac{12400}{6000} = \frac{124}{60} = 2.06 \checkmark$$

$$E_4 = \frac{12400}{7000} = 1.8 \times$$

$$I = n (E_{\text{photo}})$$

$$\text{watt} \quad \text{eV} \times 1.6 \times 10^{-19}$$

$$1.5 \times 10^{-3} = n [3.1 \times 1.6 \times 10^{-19}]$$

$$\frac{1}{3.2} \times 10^{16}$$

$$= 0.33 \times 10^{16} = n$$

$$= 3.3 \times 10^{15} = n_1$$

$$n = n_1 + n_2 + n_3$$

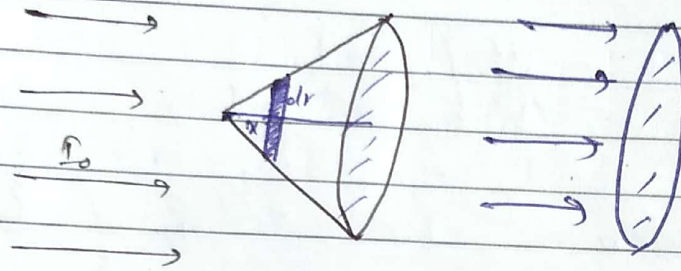
$$n_1 = 3.3 \times 10^{15} \times 10^{-4}$$

$$= 3.3 \times 10^{15} \times 10^{-4}$$

Threshold
for

because in
 m^2 .

Q. In the given diagram find force on the conical surface if absorptivity of cone is one

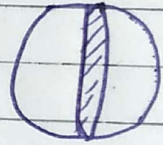


Valid only for

$$F = \frac{I_0 A}{c}$$

A = Area of circle

$$F = \frac{I_0 \pi R^2}{c}$$



$$F = \frac{I \pi R^2}{c}$$

If light is Reflected

$$dF = \frac{2I A \cos^2 \alpha}{c}$$

$$\alpha = (90^\circ - \theta)$$

$$A = (2\pi r) \frac{dr}{\sin \theta}$$

$$dF = 2I \frac{2\pi r dr}{c \sin \alpha} \sin^2 \alpha$$

$$m = 9.1 \times 10^{-31}$$

* De-broglie :

Acc. to de broglie, if light has dual nature i.e particle nature and wave nature it implies that particles are also have dual nature.

De Broglie suggested that wave is associated with each metal and the associated wave is known as metal wave.

$$p = \frac{h}{\lambda}$$

$$\lambda = \frac{h}{p}$$

$$\lambda_{m.w} = \frac{h}{p} = \frac{h}{mv} = \frac{h}{\sqrt{2m \text{K.E}}} = \frac{h}{\sqrt{2mq\Delta V}}$$

↓ Kinetic Energy
 ↓ Accelerating Potential

For $e^- \Rightarrow$ $\lambda = \sqrt{\frac{150.1}{V}}$ in $\text{\AA} \quad \text{or} \quad \frac{12.27}{\sqrt{V}} \text{\AA}$

De wave length

$$\text{K.E of molecules} \Rightarrow \text{K.E} \Rightarrow \frac{3}{2} \text{K.T}$$

$$\lambda_{gas} = \frac{h}{\sqrt{2m \frac{3}{2} KT}} = \frac{h}{\sqrt{3mKT}}$$

T is in kelvin

BT 12/15

Que. An e⁻ is Accelerated by potential diff. of 50 v. find de-Broglie wave length associated with it

Ans: (13)

$$\lambda = \sqrt{\frac{150.1}{50}} = \sqrt{3} \text{ \AA}$$

$$= \sqrt{3}$$

Q. Find the Ratio of de-Broglie. Wave length of Hydrogen of molecular of Hydrogen and Helium which are at temp. 27°C and 127°C

= $\sqrt{\frac{8}{3}}$

$$\lambda_{gas} = \frac{6.6 \times 10^{-34}}{\sqrt{3 \times 1 \times 300 \times}}$$

$$\lambda = \frac{\sqrt{2 \times 400}}{\sqrt{1 \times 300}} = \sqrt{\frac{8}{3}}$$

Q. Find the de Broglie wave length of a ball of mass 10g having a velocity of 10m/s.

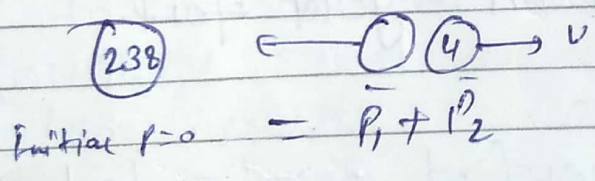
Ans: $\lambda = \frac{h}{mv} = \frac{h}{10 \times 10}$

$$= \frac{h}{0.01 \times 10}$$

Q. A Uranium nucleus decay into α -particle and its daughter nuclei. Find the ratio of de-Broglie wave length of the emitted α particle and daughter nuclei.

Ans: $U_{\text{mass}} = 238, \quad \alpha_{\text{mass}} = 4, \quad \lambda = \frac{h}{P}$

$$P = \frac{h}{\lambda} = \frac{h}{P} \quad P =$$



SBG STUDY